

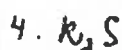
52

2. KNO_3

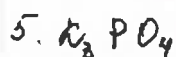
$$\omega(B-Ba) = \frac{39.1}{101} \approx 0,4.$$

~~318~~

$$\omega(k) = \frac{39.7}{85} \approx 0,5.$$



$$\omega(k) = \frac{39 \cdot 2}{110} = 0,7$$



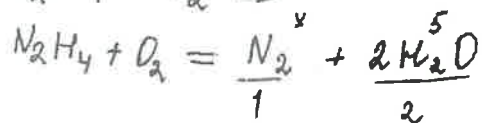
$$w(k) = \frac{39.3}{212} \approx 0,6.$$

Omber: 1243

~~54.~~ 53.~~1. $\text{NH}_3 \rightarrow \text{N}_2 + 3\text{H}_2\text{O}$~~ 

$$2.9 \text{ (H}_2\text{O)} = \frac{90.2}{18.2/\text{моль}} = 5 \text{ моль.}$$

$$M(H_2O) = 1 \cdot 2 + 16 = 18 \text{ г/моль.}$$



$$D(N_2) = \frac{5}{2} = 2,5 \text{ моль.}$$

$$\textcircled{d} V(N_2) = 2 \cdot V_M = 2,5 \cdot 22,4 = 56 \text{ (ummol)}$$

Omzet: 56 u.

NL - 0

№ 2 - 100

N3 - 100

NH-105

N5 - 0

18020 - 308 ymf.

Mail money

105

105

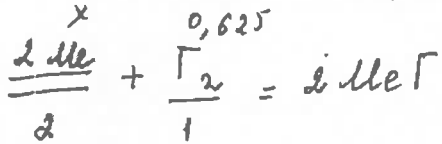
Tagh - amo $O_2, H_2, N_2 (\Gamma_2)$

$Me^+ \quad \Gamma^-$

1 : 1

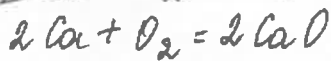
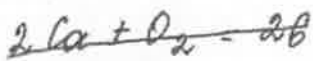


$$\gamma(\text{tagh}) = \frac{V}{V_M} = \frac{14}{22,4} = 0,625 \text{ mol}$$



$$\gamma(Me) = 0,625 \cdot 2 = 1,25 \text{ mol}$$

$$M(Me) = \frac{m}{\gamma} = \frac{50}{1,25} = 40 \text{ g/mol}$$



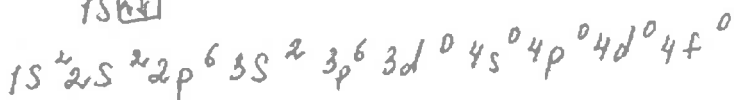
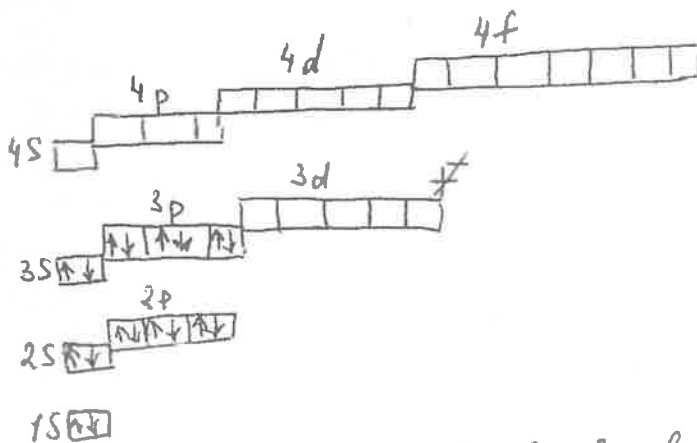
Ca: O

1 : 1

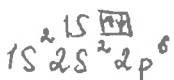
~~Ca, O₂, CaO~~ Dikern: Ca, O₂, CaO.



Ca



O₂



100